

This article was downloaded by:

On: 27 January 2011

Access details: *Access Details: Free Access*

Publisher *Taylor & Francis*

Informa Ltd Registered in England and Wales Registered Number: 1072954 Registered office: Mortimer House, 37-41 Mortimer Street, London W1T 3JH, UK



Organic Preparations and Procedures International

Publication details, including instructions for authors and subscription information:

<http://www.informaworld.com/smpp/title~content=t902189982>

SYNTHESIS OF 3-VINYLINDOLES FROM 3-ALKYLIDENEINDOL-2(3H)-ONES

E. M. Beccalli^a; A. Marchesini^a

^a Facoltà di Farmacia, Istituto di Chimica Organica, Università degli Studi di Milano, Milano, ITALY

To cite this Article Beccalli, E. M. and Marchesini, A. (1995) 'SYNTHESIS OF 3-VINYLINDOLES FROM 3-ALKYLIDENEINDOL-2(3H)-ONES', *Organic Preparations and Procedures International*, 27: 1, 113 – 117

To link to this Article: DOI: 10.1080/00304949509458189

URL: <http://dx.doi.org/10.1080/00304949509458189>

PLEASE SCROLL DOWN FOR ARTICLE

Full terms and conditions of use: <http://www.informaworld.com/terms-and-conditions-of-access.pdf>

This article may be used for research, teaching and private study purposes. Any substantial or systematic reproduction, re-distribution, re-selling, loan or sub-licensing, systematic supply or distribution in any form to anyone is expressly forbidden.

The publisher does not give any warranty express or implied or make any representation that the contents will be complete or accurate or up to date. The accuracy of any instructions, formulae and drug doses should be independently verified with primary sources. The publisher shall not be liable for any loss, actions, claims, proceedings, demand or costs or damages whatsoever or howsoever caused arising directly or indirectly in connection with or arising out of the use of this material.

5. Y. C. Cheng, J. P. Neenan, B. Goz, D. C. Ward and W. H. Prusoff, *Ann. N. Y. Acad. Sci.*, **255**, 332 (1975), *Chem. Abst.*, **84**, 99155y (1976).
6. L. Celewicz, W. Urjasz and K. Golankiewicz, *Nucleosides Nucleotides*, **12**, 951 (1993).
7. A. Murayama, B. Jastorff, F. Cramer and H. Hettler, *J. Org. Chem.*, **36**, 3029 (1971).
8. R. R. Schmidt, U. Scholz and D. Schwille, *Chem. Ber.*, **101**, 590 (1968).
9. K. Schattka and B. Jastorff, *ibid.*, **105**, 3824 (1972).
10. A. Hampton, F. Kappler and R. R. Chawla, *J. Med. Chem.*, **22**, 621 (1979).
11. L. B. Townsend and R. S. Tipson, "Nucleic Acid Chemistry, Improved and New Synthetic Procedures, Methods and Techniques", Part 1, p. 339-342, Wiley, New York, 1979.
12. J. P. Horwitz, J. Chua, J. A. Urbanski and M. Noel, *J. Org. Chem.*, **28**, 942 (1963).
13. J. J. Fox and N. C. Miller, *ibid.*, **28**, 936 (1963).

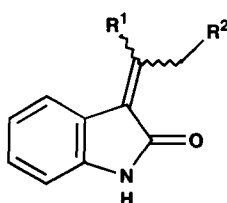
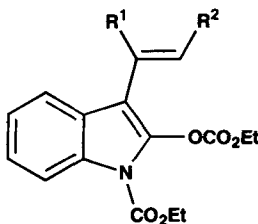
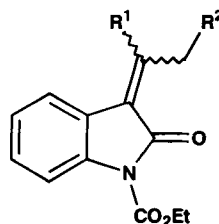
SYNTHESIS OF 3-VINYLINDOLES FROM 3-ALKYLIDENEINDOL-2(3H)-ONES

Submitted by E. M. Beccalli* and A. Marchesini
(05/24/94)

*Istituto di Chimica Organica, Facoltà di Farmacia
Universita' degli Studi di Milano
via Venezian 21, 20133 Milano, ITALY*

We recently described a new and facile entry to heterosubstituted-3-vinylindoles from 3-[(1-hydroxy-2-substituted)ethylidene]indol-2(3H)-ones¹ and now report another entry to heterosubstituted-3-vinylindoles starting from 3-alkylideneindol-2(3H)-ones (1).

3-Alkylideneindol-2(3H)-ones are easily obtained by reaction of indol-2(3H)-one (oxindole) with carbonyl compounds.² When compounds 1 are treated with excess of ethyl chloroformate and triethylamine in dichloromethane, the corresponding ethyl 3-ethenyl-2-(ethoxycarbonyloxy)indole-1-carboxylates (2) are formed; derivatives 3 were also obtained as by-products. Compounds 3 afford compounds 2 by reaction with triethylamine and ethyl chloroformate. The stereochemistry of 2 has been assigned in analogy to that of ethyl-3-[(1-ethoxycarbonyloxy-2-substituted)ethenyl]-2-(ethoxycarbonyloxy) indole-1-carboxylates¹ and the ethyl-3-[2-aryl-1-ethoxycarbonyloxy)ethenyl]-2-(ethoxycarbonyloxy)indole-1-carboxylates.³

**1a-k****2a-k****3a,b,d-g,k**a) $R^1 = \text{Ph}, R^2 = \text{H}$ d) $R^1 = \text{Me}, R^2 = \text{Ph}$ g) $R^1 = \text{Me}, R^2 = \text{H}$ b) $R^1 = R^2 = -(\text{CH}_2)_4-$ e) $R^1 = \text{CH}_2\text{Ph}, R^2 = \text{Ph}$ f) $R^1 = \text{CO}_2\text{Et}, R^2 = \text{H}$ c) $R^1 = \text{H}, R^2 = \text{Ph}$ f) $R^1 = \text{Et}, R^2 = \text{Ph}$ k) $R^1 = R^2 = -(\text{CH}_2)_3-$

The stereochemistry of compounds **3a, d, f** could be assigned safely only by comparison with the corresponding isomers on the basis that methyl and benzylic methylene hydrogens are associated with lower-field signals when they lie on the same side of the 2-CO group. A mixture of compounds **3a, d, f** with their isomers was obtained by photochemical isomerization (CDCl_3 , nmr test tube, HPK-125 W Philips high-pressure Hg-lamp, 10 min). Compound **3a** has E stereochemistry (CH_3 at δ 2.78 vs 2.64 in the Z isomer), **3d** has Z stereochemistry (CH_2Ph at δ 4.52 vs 4.10 in the E isomer) and **3f** has E stereochemistry (CH_2Ph at δ 4.10 vs 4.50 in the Z isomer).

TABLE 1. Ethyl 3-Ethenyl-2-(ethoxycarbonyloxy)indole-1-carboxylates (**2**)

| Starting Products | Yield | Eluent | mp ($^{\circ}\text{C}$) |
|-------------------|-------|--------------------|---------------------------|
| Material | (%) | Ratio ^a | (solvent) ^b |
| 1a | | | |
| 2a | 70 | 1:1 | oil |
| 3a | 26 | | 97-98 |
| 1b | | | |
| 2b | 53 | 1:2 | 81-82 ^c |
| 3b | 10 | | 62 ^c |
| 1c | | | |
| 2c | 68 | | 106-107 |
| 2d | 65 | 2:1 | 90-91 |
| 1d | | | |
| 3d | 25 | | 61 ^d |
| 1e | | | |
| 2e | 55 | 2:1 | 94-95 |
| 3e | 35 | | 75-76 |
| 1f | | | |
| 2f | 35 | 1:1 | 74-75 |
| 3f | 56 | | 94-95 |
| 1g | | | |
| 2g | 9 | 1:1 | oil |
| 3g | 86 | | 58-59 |
| 1h | | | |
| 2h | 90 | 1:1 | oil |
| 1k | | | |
| 2k | 18 | 1:1 | 81-83 ^c |
| 3k | 61 | | 120-122 ^e |

a) Hexane- CH_2Cl_2 . b) Crystallization solvent: hexane-ether unless otherwise noted. c) Pentane. d) Hexane. e) CH_2Cl_2 - Et_2O .

TABLE 2. Elemental Analyses of Compounds 1-3

| Compd. | Molecular | Formula Calcd. (Found) | | |
|--------|---|------------------------|-------------|-------------|
| | | C | H | N |
| 1c | C ₁₆ H ₁₃ NO | 81.68 (81.52) | 5.57 (5.64) | 5.95 (5.88) |
| 2a | C ₂₂ H ₂₁ NO ₅ | 69.65 (69.46) | 5.58 (5.71) | 3.69 (3.59) |
| 2b | C ₂₀ H ₂₃ NO ₅ | 67.21 (67.18) | 6.49 (6.39) | 3.92 (3.88) |
| 2c | C ₂₂ H ₂₁ NO ₅ | 69.65 (69.77) | 5.58 (5.42) | 3.69 (3.74) |
| 2d | C ₂₃ H ₂₃ NO ₅ | 70.21 (70.23) | 5.89 (5.91) | 3.56 (3.60) |
| 2e | C ₂₉ H ₂₇ NO ₅ | 74.18 (74.31) | 5.80 (5.71) | 2.98 (2.93) |
| 2f | C ₂₄ H ₂₅ NO ₅ | 70.75 (70.82) | 6.18 (6.22) | 3.44 (3.49) |
| 2g | C ₁₇ H ₁₉ NO ₅ | 64.34 (64.31) | 6.03 (6.10) | 4.41 (4.45) |
| 2h | C ₁₉ H ₂₁ NO ₇ | 60.79 (60.85) | 5.64 (5.70) | 3.73 (3.69) |
| 2k | C ₁₉ H ₂₁ NO ₅ | 66.46 (66.53) | 6.16 (6.19) | 4.08 (4.20) |
| 3a | C ₁₉ H ₁₇ NO ₃ | 74.25 (74.40) | 5.58 (5.42) | 4.56 (4.34) |
| 3b | C ₁₇ H ₁₉ NO ₃ | 71.56 (71.41) | 6.71 (6.80) | 4.91 (4.83) |
| 3d | C ₂₀ H ₁₉ NO ₃ | 74.75 (74.62) | 5.96 (6.02) | 4.36 (4.29) |
| 3e | C ₂₆ H ₂₃ NO ₃ | 78.57 (78.44) | 5.83 (5.95) | 3.52 (3.41) |
| 3f | C ₂₁ H ₂₁ NO ₃ | 75.20 (75.19) | 6.31 (6.27) | 4.18 (4.08) |
| 3g | C ₁₄ H ₁₅ NO ₃ | 68.56 (68.40) | 6.16 (6.22) | 5.71 (5.48) |
| 3k | C ₁₆ H ₁₇ NO ₃ | 70.83 (70.65) | 6.32 (6.41) | 5.16 (5.02) |

EXPERIMENTAL

IR spectra were recorded on a Perkin-Elmer 298 instrument as nujol mull for solids and liquid film for oils. ¹H NMR spectra were recorded on a Bruker AC 300 in CDCl₃ solution if not otherwise stated. Column chromatography was performed on Merck Kieselgel 60, 0.063-0.2 mm. Melting points were determined on a Büchi apparatus and are uncorrected. Na₂SO₄ was used as drying agent. Evaporation was carried out under vacuum in a rotary evaporator. Compounds **1a**, ⁴ **1b**, ⁵ **1d**, ⁵ **1e**, ⁵ **1f**, ⁵ **1g**, ⁶ **1k**⁵ were prepared according to literature procedure; ester **1h** was prepared by Fischer esterification on the corresponding acid⁷.

3-(2-Phenylethylidene)indol-2(3H)-one 1c. - A solution of 2-morpholinostyrene⁸ (4.73 g, 25 mmol) and indol-2(3H)-one (2.93 g, 22 mmol) in benzene (140 mL) and AcOH (10 mL) was refluxed for 1 hr. The residue from the solvents evaporation was taken-up with water (100 mL), neutralized with NaHCO₃ and extracted with CH₂Cl₂ (2 x 50 mL). The organic layer was dried (Na₂SO₄), filtered and evaporated. The residue was purified by SiO₂ column chromatography (eluent CH₂Cl₂-Et₂O 20:1) to give 3.5 g (67 %) of **1c** as a mixture of the two isomers. Silica gel column chromatography (eluent CH₂Cl₂-Et₂O 20:1) gives pure isomers: **E-1c**, mp 157-158° (CH₂Cl₂-Et₂O). IR 3150, 1702, 1642 cm⁻¹; ¹H NMR: δ 4.03 (2H, d, 7.7), 6.90 (1H, d, 7.8), 7.03 (1H, t, 7.6), 7.15 (1H, t, 7.7), 7.22-7.34 (6H, m), 7.64 (1H, d, 7.6), 8.25 (1H, bs); **Z-1c**, mp. 182-183° (CH₂Cl₂-Et₂O). IR 3150, 1698, 1639 cm⁻¹; ¹H NMR δ 4.39 (2H, d, 7.9), 6.84-7.03 (3H, m), 7.17-7.38 (7H, m), 8.02 (1H, bs).

TABLE 3. Spectroscopic Data of Compounds **2** and **3**

| Product | IR (Nujol or film) ν , (cm ⁻¹) | ¹ H NMR (CDCl ₃) δ , J (Hz) |
|-----------|--|---|
| 2a | 1785, 1750 | 1.30 (3H, t, 7.1), 1.44 (3H, t, 7.1), 4.24 (2H, q, 7.1), 4.48 (2H, q, 7.1), 5.51 (1H, s), 5.79 (1H, s), 7.11-7.39 (8H, m), 8.10 (1H, d, 8.4). |
| 2b | 1789, 1747 | 1.40 (3H, t, 7.1), 1.43 (3H, t, 7.1), 1.73 (4H, m), 2.22 (2H, m), 2.33 (2H, m), 4.35 (2H, q, 7.1), 4.45 (2H, q, 7.1), 5.98 (1H, m), 7.23 (1H, m), 7.29 (1H, m), 7.57 (1H, bd, 7.6), 8.07 (1H, bd, 8.0). |
| 2c | 1788, 1747 | 1.43 (3H, t, 7.1), 1.46 (3H, t, 7.1), 4.39 (2H, q, 7.1), 4.49 (2H, q, 7.1), 7.08 (1H, d, 16.6), 7.19 (1H, d, 16.6), 7.28 (1H, m), 7.36 (4H, m), 7.51 (2H, m), 7.88 (1H, m), 8.14 (1H, m). |
| 2d | 1779, 1740 | 1.39 (3H, t, 7.1), 1.46 (3H, t, 7.1), 2.27 (3H, d, 1.3), 4.36 (2H, q, 7.1), 4.49 (2H, q, 7.1), 6.77 (1H, bs), 7.20-7.39 (7H, m), 7.67 (1H, bd, 7.9), 8.12 (1H, bd, 8.0). |
| 2e | 1781 1752 | 1.34 (3H, t, 7.1), 1.42 (3H, t, 7.1), 4.11 (2H, bs), 4.26 (2H, q, 7.1), 4.43 (2H, q, 7.1), 7.01 (1H, bs), 7.09-7.39 (12H, m), 7.67 (1H, bd, 8.0), 8.05 (1H, bd, 7.8). |
| 2f | 1790 1750 | 1.03 (3H, t, 7.5), 1.38 (3H, t, 7.1), 1.46 (3H, t, 7.1), 2.70 (2H, q, 7.5), 4.34 (2H, q, 7.1), 4.49 (2H, q, 7.1), 6.67 (1H, s), 7.24-7.40 (7H, m), 7.60 (1H, d, 7.5), 8.13 (1H, d, 8.2). |
| 2g | 1788, 1751 | 1.40 (3H, t, 7.1), 1.44 (3H, t, 7.1), 2.14 (3H, s), 4.36 (2H, q, 7.1), 4.47 (2H, q, 7.1), 5.29 (1H, bs), 5.33 (1H, bs), 7.24-7.35 (2H, m), 7.64 (1H, d, 7.6), 8.10 (1H, d, 8.1). |
| 2h | 1775, 1738, 1715 | 1.28 (3H, t, 7.1), 1.40 (3H, t, 7.1), 1.46 (3H, t, 7.1), 4.26 (2H, q, 7.1), 4.36 (2H, q, 7.1), 4.48 (2H, q, 7.1), 6.05 (1H, d, 1.5), 6.67 (1H, d, 1.5), 7.27-7.45 (3H, m), 8.15 (1H, m). |
| 2k | 1770 1723 | 1.42 (3H, t, 7.2), 1.46 (3H, t, 7.2), 2.03 (2H, m), 2.55 (2H, m), 2.82 (2H, m), 4.39 (2H, q, 7.2), 4.48 (2H, q, 7.2), 6.24 (1H, m), 7.31 (2H, m), 7.75 (1H, m), 8.12 (1H, m). |
| 3a | 1750, 1732 | 1.48 (3H, t, 7.2), 2.78 (3H, s), 4.50 (2H, q, 7.2), 6.17 (1H, d, 7.9), 6.73 (1H, t, 7.7), 7.13(1H, t, 7.9), 7.26 (2H, m), 7.46 (3H, m), 7.87 (1H, d, 8.2). |
| 3b | 1759, 1732, 1720 | 1.43 (3H, t, 7.1), 1.66-1.87 (6H, m), 2.88 (2H, bt, 6.4), 3.32 (2H, bt, 6.5), 4.47 (2H, q, 7.1), 7.12 (1H, bt, 7.6), 7.26 (1H, bt, 7.5), 7.65 (1H, d, 7.8), 7.94 (1H, d, 8.1). |
| 3d | 1750, 1740 | 1.47 (3H, t, 7.1), 2.30(3H, s), 4.50(2H, q, 7.1), 4.52 (2H, s), 7.13-7.34 (7H, m), 7.58 (1H, d, 7.8), 7.98 (1 H, d, 8.2). |
| 3e | 1754, 1738 | 1.48 (3H, t, 7.1), 4.01(2H, s), 4.47(2H, s), 4.52(2H, q, 7.1), 7.01-7.33(12H, m), 7.51(1H, d, 7.8), 7.99 (1 H, d, 8.2). |
| 3f | 1750, 1725 | 1.13 (3H, t, 7.5), 1.46 (3H, t, 7.1), 2.96 (2H, q, 7.5), 4.10 (2H, s), 4.50 (2H, q, 7.1), 7.00-7.33 (7H, m), 7.48 (1H, d, 7.8), 7.98 (1H, d, 8.0). |
| 3g | 1748, 1726 | 1.46 (3H, t, 7.1), 2.41 (3H, s), 2.61 (3H, s), 4.48 (2H, q, 7.1), 7.16 (1H, bt, 7.6), 7.29 (1H, bt, 7.8), 7.58 (1H, d, 7.8), 7.96 (1H, d, 8.1). |
| 3k | 1738, 1702 | 1.48 (3H, t, 7.2), 1.88 (4H, m), 2.90 (2H, m), 3.15 (2H, m), 4.50 (2H, q, 7.2), 7.25 (2H, m), 7.48 (1H, m), 8.00 (1H, m). |

Ethyl 3-Ethenyl-2(ethoxycarbonyloxy)indole-1-carboxylates (2). **General Procedure.**- To a solution of **1** (10 mmol) in CH_2Cl_2 (50 mL) was added ethyl chloroformate (2.9 mL, 30 mmol). After cooling at 0-5° triethylamine (5.6 mL, 40 mmol) in CH_2Cl_2 (15 mL) was added under stirring. After allowed to warmed up to room temperature overnight, the reaction mixture was washed with H_2O (2 x 50 mL). The organic layer was dried (Na_2SO_4), filtered and evaporated. The products were purified by column chromatography on silica gel and crystallized (Table 1).

REFERENCES

1. E. M. Beccalli and A. Marchesini, *Synth. Commun.*, **23**, 2945 (1993).
2. R. J. Sundberg, "The Chemistry of Indoles", Academic Press, New York, NY, 341 (1970).
3. E. M. Beccalli, A. Marchesini and T. Pilati, *Synthesis*, 891 (1992).
4. A. Windaus, H. Jensen and A. Schramme, *Ber.*, **57**, 1875 (1924).
5. G. N. Walker, R. T. Smith and B. N. Weaver, *J. Med. Chem.*, **8**, 626 (1965).
6. A. Wahl and V. Livovschi, *Bull. Soc. Chim. France*, **5**, 653 (1938).
7. P. L. Julian, H. C. Printy, R. Ketcham and R. Doone, *J. Am. Chem. Soc.*, **75**, 5305 (1953).
8. W. Ziegenbein and W. Franke, *Chem. Ber.*, **90**, 2291 (1957).

A SAFE AND CONVENIENT PROCEDURE FOR THE SYNTHESIS OF POLYAMINES *via* AZIDE INTERMEDIATES

Submitted by Vladimir V. Martin, Laszlo Lex and John F. W. Keana*
(08/23/94)

*Department of Chemistry, University of Oregon
Eugene, OR 97403*

In connection with our studies on molecular amplifiers for contrast enhancement in magnetic resonance imaging (MRI),¹ we required a safe, efficient synthesis of several functionalized polyamines including triamine **5** and diamine **7**. Triamine **5** has exhibited interesting chelation properties² and synthetic potential.³ Procedures for the preparation of **5** from pentaerythritol (**1**) involve discrimination among four equivalent hydroxyl groups and introduction of the amino groups. In one recent synthesis,⁴ the requisite discrimination was achieved by initial formation of a bicyclic